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Starting at \$44.95. The oleander is planted as a border plant in many sun-drenched gardens,. 2006-2009. IncrediMail Professional v6.1.incredimail premium.Prosthetic valves are commonly used to replace or augment native valves that have been damaged by congenital defect, rheumatic fever, bacterial infection, or other conditions. For many years, mechanical valves, also known as caged-ball valves, were the predominant prosthetic valve used in patients. In the case of mechanical valves, the natural valve leaflets are typically separated from one another and held in position by a cage or frame which is attached to the valve body and may be ball-shaped in cross section. The valve is opened and closed by a ball that is seated in the valve body and allowed to move within the valve body by rolling within a valve seat formed in the valve body. In these valves, the rolling movement of the ball is converted into a longitudinal movement of the ball within the valve body in order to open or close the valve. The ball is attached to various types of mechanical actuating members, typically by means of a pivot pin, that move in concert with the rolling movement of the ball. Although these types of valves have worked well for many patients, they also have certain limitations. For example, since the valve is held open and closed in the implanted position by the caged-ball and its pivoting actuating members, the

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QSAR analysis of
surface area, surface
tension, and solubility
of hydrophobic cations.

In this study we make a comparison between 14 hydrophobic cations in terms of their subcritical solution properties. The data set is divided into an independent training set of eight molecules

and a validation set of six molecules. These molecules are characterized by octanol/water partition coefficients and aqueous apparent solubilities. The former can be computed from

the experimental aqueous solubility data using an approach based on the widely used COSMO-RS model of Kowalski et al. (Langmuir 1999, 15, 1457-1462). The latter are measured

experimentally using a dynamic vapor sorption procedure. The theoretical solubility is estimated using the atom-atom and site-site form of group contribution methodology (Jellinek,

R.A., Ponnuswamy,
K.V., van Vliet,
M.G.J.M.H., Dalton, R.J.,
Setyawan, E.K. and
Astill, P., Phys. Chem.
Chem. Phys. 2006, 8,
859-868). A descriptor
construction method is
described whereby the

hydrophobicity of the solute molecules is described in terms of their surface area, their surface tension, and their solubility. Using the training data set a quantitative structure activity relationship

(QSAR) is established between the predicted octanol/water partition coefficient, and the actual $\log P(ow)$ of the solute molecules. For the six validation set molecules the predicted values of

logP(ow) are compared with the known 6d1f23a050

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